

Al-C-Cr (Aluminum-Carbon-Chromium)

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Schuster et al. [1980Sch] presented two isothermal sections at 1000 and 800 °C for this system. Recently, [2006Hal] determined the thermodynamic properties of the ternary compound Cr_2AlC and computed a liquidus projection and two isothermal sections at 1400 and 1000 °C.

$\beta\text{Cr}_4\text{Al}_9$, $\gamma\text{Cr}_4\text{Al}_9$, $\alpha\text{Cr}_5\text{Al}_8$, and $\beta\text{Cr}_5\text{Al}_8$, with no clear phase boundaries between them [2000Mah]. The C-Cr phase diagram [1990Ven] depicts three well-established compounds Cr_{23}C_6 ($D8_4$ -type cubic), Cr_7C_3 ($D10_1$ -type orthorhombic), and Cr_3C_2 ($D5_{10}$ -type orthorhombic).

Binary Systems

In the Al-C system [1991Har], the stoichiometric compound Al_4C_3 ($D7_1$ -type rhombohedral) is present. The Al-Cr phase diagram by [2000Mah] includes a thermodynamic assessment and depicts a number of intermediate phases with significant ranges of homogeneity: CrAl_7 ($V_7\text{Al}_{45}$ -type monoclinic), $\text{Cr}_2\text{Al}_{11}$ (CrAl_5 -type monoclinic), CrAl_4 (monoclinic), Cr_2Al (MoSi_2 -type tetragonal), and an unconfirmed low-temperature phase X at 75 at.% Cr. Between 30 and 41 at.% Cr, five phases have been reported: $\alpha\text{Cr}_4\text{Al}_9$,

Ternary Phase Equilibria

[1980Sch] determined the crystal structure of the ternary compound Cr_2AlC (denoted τ here; hexagonal, $P6_3/mmc$, 8 atoms/cell, $a = 0.2866$ nm and $c = 1.282$ nm). Employing ab initio calculations, [2006Hal] estimated the energy of formation of Cr_2AlC . The incongruent melting temperature determined from the not-so-precise DTA measurement was between 1460 and 1495 °C and agrees satisfactorily with the calculated value of 1498 °C. [2006Hal] accepted the binary descriptions of [1995Gro] (Al-C), [1992Lee] (Cr-C)

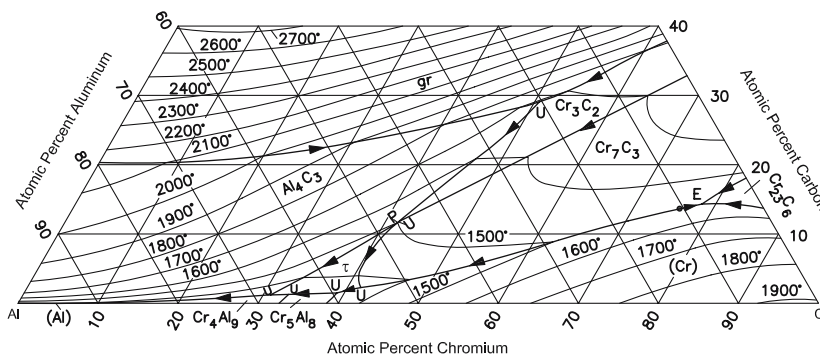


Fig. 1 Al-C-Cr computed liquidus projection [2006Hal]

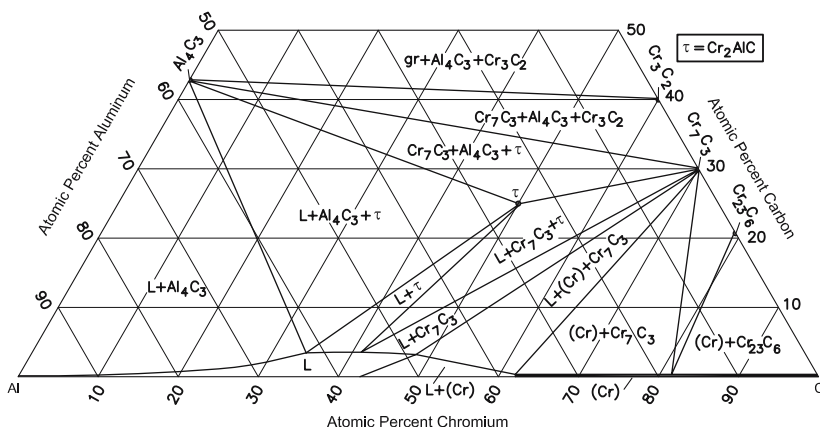


Fig. 2 Al-C-Cr computed isothermal section at 1400 °C [2006Hal]. Narrow two-phase regions are omitted

